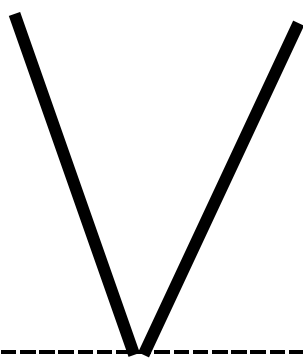
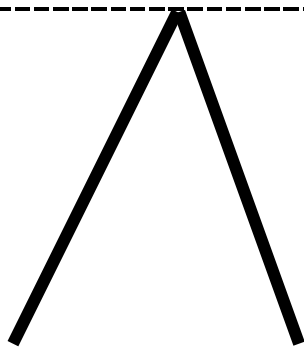


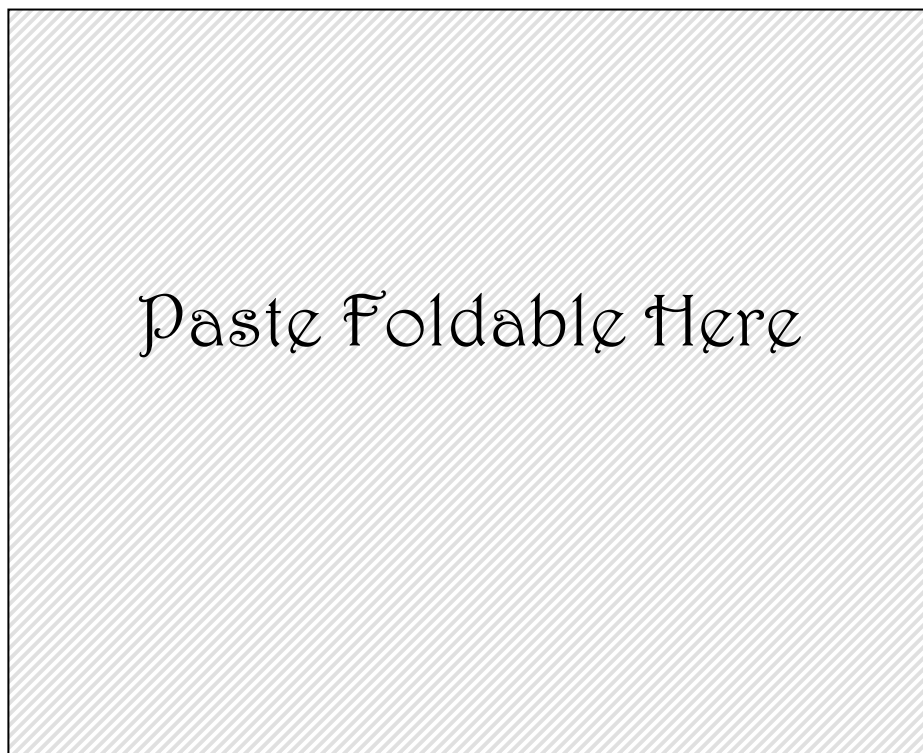
<ul style="list-style-type: none"> ➤ A stands for: ➤ Definition: ➤ Purpose: 	<ul style="list-style-type: none"> ➤ X stands for: ➤ Text Formatting: First letter always following letters always 	<ul style="list-style-type: none"> ➤ Charge tells: ➤ Relationship to subatomic particles: Negative charge: Neutral charge: Positive charge:
<ul style="list-style-type: none"> ➤ Relationship to subatomic particles: A = # of + # of 	<p><u>EXAMPLES:</u></p>	
<ul style="list-style-type: none"> ➤ Relationship to subatomic particles: Z = # of ➤ Where can the value for Z be found? 	<p>¹³C</p> <p>Protons = _____ Atomic # = _____</p> <p>Neutrons = _____ Mass # = _____</p> <p>Electrons = _____ Charge = _____</p> <p>Name = _____</p>	
<ul style="list-style-type: none"> ➤ Z stands for: ➤ Definition: ➤ Purpose: 	<p>²³⁸U</p> <p>Protons = _____ Atomic # = _____</p> <p>Neutrons = _____ Mass # = _____</p> <p>Electrons = _____ Charge = _____</p> <p>Name = _____</p>	

charge



Atoms by the Numbers

Isotopic Notation



M.

- A.

N.

Battle of the I words...

Ions

Isotopes

Average Atomic Mass

*Either: Paste Back of
folded Average
Atomic Mass Activity
or Make a Pocket
here*